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TEKs: 6.5A – know
that an element is a
pure substance

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represented by
chemical symbols.

6.5B – recognize
that a limited
number of the many
known elements
comprise the
largest portion of
solid Earth, living
matter, oceans, and
the atmosphere.

6.5C – differentiate
between elements
and compounds on

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the most basic
level. 6.5D ...

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Unit 3: Chemistry Matter and Elements

In Class:

"Chemistry Unit
test review packet"
(see below)

Homework: Finish
review packet and
Study for test

Tuesday 11.18.14.

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Thursday 11.13.14.

In Class: Mrs.
Cushing was out.

You read an article
on Nucleic Acids
and took notes on
the reading, you
may have begun the
Chemistry Review
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Homework: Study
for Test Tuesday
11.18.14

Wednesday

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11.12.14. In Class:
We reviewed all of
the ...

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TEKs: 6.5A – know that an element is a pure substance represented by chemical symbols.

6.5B – recognize that a limited number of the many known elements comprise the largest portion of solid Earth, living matter, oceans, and the atmosphere.

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11.Highlight the non-

metals on the

Periodic Table

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REVIEW Atoms:

The Building Blocks

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2 SHORT ANSWER
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Answer the
following questions

in the space
provided. 1. In
cathode-ray tubes,
the cathode ray is
emitted from the
negative electrode,
which is called the
...

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Atomic Structure
and the Periodic
Table Test Review
answers to

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atoms 2. different

3. compounds 4..

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Agenda . Elements

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Periodic Table. 11..

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TEKs: 65A – know
that an element is a
pure substance
represented by
chemical symbols
65B – recognize
that a limited
number of the many
known elements
comprise the
largest portion of

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the associate to

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3. The elemental
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The main function of carbohydrates. 6.

An example of a polysaccharide. 8.

The basic building block/unit used to make polymers. 9.

An example of a disaccharide. 11.

An example of a monosaccharide.

Down. 1. The polymer of carbohydrates. 2.

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BIOLOGY REVIEW

PACKET - Michigan

State University

Unit 3: Periodic

Table Practice

Packet ____ 1. I can

classify elements as

metals, nonmetals,

or metalloids based

on their placement

on the Periodic

Table. Classify each

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of the following
elements as metals
(M), nonmetals
(NM), or metalloids
(MTLD).

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