

Stoichiometry Chapter Test B Answers

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CHAPTER 9 REVIEW Stoichiometry SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. b The coefficients in a chemical equation represent the (a) masses in grams of all reactants and products. (b) relative number of moles of reactants and products. (c) number of atoms of each element in each compound in a reaction.

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Chapter Test B Stoichiometry Answers Chapter Test B Chapter: Stoichiometry PART I Inthe space provided, write the letter ofthe term or phrase that best completes each statement or bestanswers each question. 1. Knowingthe mole ratio ofa reactantandproductin a chemical reaction would allowyoutodetermine a. the energyreleased inthe reaction.

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Remember it is a MC test, use the answers ... Practice Test Ch3 Stoichiometry (page 2 of 2) 19. The mass of element X found in 1.00 mole of each of four different compounds is 28.0 g, 42.0 g, 56.0 g, and 70 g, respectively. The possible atomic weight of X is a. 8.00 b. 14.0

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Modern Chemistry 69 Chapter Test Chapter: Chemical Equations and Reactions PART I In the space provided, write the letter of the term or phrase that best com-pletes each statement or best answers each question. ____ 1. The production of a slightly soluble solid compound in a double-displacement reaction results in the formation of a a. gas. b ...