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An electron is contained in a three-dimensional rectangular box given by $0 \leq x \leq a, 0 \leq y \leq b,$ and $0 \leq z \leq c$. The solutions of the Schrödinger equation are specified by the quantum numbers $n_x, n_y,$ and n_z . Recall that the normalized wave function is...with energy...where $n_x, n_y,$ and n_z are positive integers.

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$r,$ and $|(l+1) - m - 2 - 1/4,$ which is to say, $|2+ l +1/4 = - m^2,$ or $(l+1/2)^2= - m^2,$ or $l = |m - | - 1/2.$ [Our present equation depends only on $- m^2,$ and hence is the same for either sign, but the solution to Problem 4.39 assumed $l+1/2 = 0$ (else u is not normalizable), so we need $|m|$ here.] Quoting 4.39: $E = (j.$

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