

## Physical Science Thermal Energy Chapter Test Answers

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thermodynamics. the study of energy transformations that occur in a collection of matter. third law of thermodynamics. the entropy of a perfect crystal at absolute zero is zero.  $Q = m \cdot TC$ . Equation for thermal Energy. Conduction. a spoon getting hot in a hot bowl of soup is an example. Convection.

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thermal energy. a measure of the sum of all the kinetic and potential energy of all molecules. temperature mass. The higher the (t)\_\_\_\_\_ the more thermal energy. The more (m)\_\_\_\_, the more thermal energy. change in thermal energy = (mass)(specific heat)(change in temperature) WORD FORM.

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Physical Science - Chapter 6 - Thermal Energy. The sum of kinetic energy and potential energy of the particles that make up a material. The average kinetic energy of the particles that make up a material. The movement of thermal energy from a warmer object to a cooler object.

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The study of the relationships between thermal energy, heat, and work. First Law of Thermodynamics. States that if the mechanical energy of a system is constant, the increase in thermal energy of that system equals the sum of the thermal energy transfers into that system and the work done on that system. Second Law of Thermodynamics.

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object increases.Because thermal energy is the total kinetic and potential energy of all the molecules in an object,the thermal energy ofthe object increases when the average kinetic energy of its molecules increases.Therefore,the thermal energy ofan object increases as its temperature increases. Thermal Energy and Mass Suppose you have a glass and a

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Ch.11 Review Physical Science 1. What is heat? Heat - Thermal energy that is moving. 2. Why do objects heat up or cool down? 3. How are heat and temperature the same? They are related but not the same... 4. Explain why heat and temperature aren ' t the same thing. 5. What units is heat measured in? 6. Explain what a calorie is? 7.

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Physical science thermal energy worksheet answers and unit study guide answer key. 8 when this substance is melting the temperature of the ice water mixture remains constant because. 1 is the transfer of thermal energy by particles moving through a fluid. Heat energy is being converted to potential energy.