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your procedures and decide how you will collect your data. Perform the virtual experiment and analyze your results. Develop a scientific argument (claim, evidence, reasoning) that answers the research question.

Inquiry Lab Spectroscopy - Mr. Palermo's Flipped Chemistry

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Mr Palermo Spectroscopy Lab Answers EXPERIMENT 8:

ATOMIC EMISSION SPECTRA Name: Instructor Date:

Section/Group: Post-Lab Report (Use the In-lab data to complete the lab report. Complete and turn in this portion along with the post-lab questions.) PARTA Gas Discharge

Tube #1 Element Hidrosep Color Wavelength (nm)

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When a beam of white light is shone through a prism, a rainbow of colors spread out to the right of the

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In this virtual lab you will: 1.Observe the bright line spectra (emission spectra) for various elements. 2.Use a flame test to

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observe the color produced when metal ions are heated.

3. Identify unknown metals ions based on the results of the flame test. Procedure: Part I. (Bright Line Spectra)

Virtual Lab Spectroscopy - Mr. Palermo's Flipped Chemistry

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The flipped classroom is not an online course. It is a way to engage students and make them more active in the learning process. Students watch a short video lecture for homework (click on the video below for an example) and take notes and complete practice examples. In the classroom these problems are evaluated by the students and teacher to address difficulties and misconceptions.

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Back to Lab. Barium Flame Test. Calcium Flame Test.  
Rubidium flame test. Potassium Flame test. Lithium Flame  
Test. Sodium Flame test. CONNECT:

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understand why the answers are what they are. Lab 6:  
Astronomical Spectroscopy A)The Prism 1. Passing light

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through a prism produces a continuous spectrum. 2. The light source is an incandescent bulb. We expect it to produce a continuous spectrum because the light source is an opaque body (a

## Lab 6: Astronomical Spectroscopy

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