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damage tissues.

~~Chapter 17~~

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Acids are named by changing

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the ending of the alkane to -oic acid. i.e. a 4 carbon acid goes from butane to butanoic acid Acid salts are named by changing the acid ending to -oate i.e the 4 carbon acid goes from butanoic acid to sodium butanoate Esters are named by changing the ending of the alkane to -oate and by using alkyl name i.e. the 4 carbon ester below is methyl butanoate (methyl for OCH₃ portion) O O Anhydrides are dehydrated esters that replace -oic acid with word anhydride Acyl ...

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17: Acids and Bases. acid.
base. alkaloids. Arrhenius.
sour taste, dissolves
metals, turns blue litmus
paper red, mor... bitter
taste, slippery feel, turns
red litmus paper blue,
more... organic bases found
in plants that are often
poisonous.

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Carboxyl, amine, hydrogen
and variable side group
Describe two forces that
help shape an amino acid
chain? Peptide bond holds
chains of amino acids
together by linking amine

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and carboxyl groups; hydrogen bonds help fold proteins into shapes and hold them together. Shape depends on amino acids, polarity and R groups.

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different proteins.

Arranging amino acids in different sequences yields the body's myriad of proteins. Two general classes of amino acids: essential and nonessential. NASM Study Guide Chapter 17 - Nutrition - The Healthy Gamer model of acids and

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•The most common strong acids are HCl, HBr, HI, HNO₃, HClO₃, HClO₄, and H₂SO₄. •Strong acids are strong electrolytes. •All strong acids ionize completely in solution. •Example: Nitric acid ionizes completely in aqueous solution. HNO₃(aq)

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+ H₂O(l) H₃O⁺(aq) + NO₃⁻(aq) • Since H⁺ and H₃O⁺ are used interchangeably, we write HNO₃

~~AP Chemistry CHAPTER 16 STUDY GUIDE Acid-Base Equilibrium~~

There are 20 amino acids, but only 4 nucleotide bases in DNA. That means it takes at least 3 nucleotides to code for one amino acid, hence a codon (three amino acids). The code is read as an RNA code, 5' to 3'.

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